

## Structure Of The Atom Answer Key|courieri font size 14 format

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Lewis structures show each atom and its position in the structure of the molecule using its chemical symbol. Lines are drawn between atoms that are bonded to one another (pairs of dots can be used instead of lines). Excess electrons that form lone pairs are represented as pairs of dots, and are placed next to the atoms. Although main group elements of the second period and beyond usually react ...

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So, a Lewis structure of water (H 2 O) would have a total of 8 electrons, one for each of the 2 hydrogen atoms and 6 for the single oxygen atom. In general, when forming bonds, electrons will be rearranged until each atom in the compound is surrounded by 8 electrons.